

## Thermodynamic Calculus §5.1

Two key problems in thermodynamics:

⇒ relate properties to each other,

e.g.  $C_p = C_v + R$ ; how about a general relationship between  $C_p = \left(\frac{\partial H}{\partial T}\right)_p$  and  $C_p = \left(\frac{\partial u}{\partial T}\right)_p$

⇒ measure them in the laboratory

From §1.4:  $n+2$  <sup>independent</sup> variables fully characterize a system  
 (e.g.  $P, V$  are not independent)  
 For one-component system, need 3 variables

Combined First + Second Laws:

$$du = Tds - PdV \quad \text{Fundamental Equation}$$

$S, V$  always "good" (independent) variables

Need one more →  $N$ , number of moles

Complete expression for  $u(S, V, N)$

$$du = \left(\frac{\partial u}{\partial S}\right)_{V, N} ds + \left(\frac{\partial u}{\partial V}\right)_{S, N} dV + \left(\frac{\partial u}{\partial N}\right)_{S, V} dN$$

$$\text{where } \left(\frac{\partial u}{\partial S}\right)_{V, N} = T \quad \left(\frac{\partial u}{\partial V}\right)_{S, N} = -P$$

$$\text{and } \left(\frac{\partial u}{\partial N}\right)_{S, V} \equiv \mu$$

$$\Rightarrow \boxed{du = Tds - PdV + \mu dN}$$

defines the chemical potential

most important equation in thermodynamics  
 fundamental Equation

What is the chemical potential?

- drives reactions / sensors / taste / smell
- measured by pH meters etc
- increases with concentration, but can be the same for different concentrations (e.g. Liq./gas equil.)

Generalization of F.E. for mixtures:

$$du = Tds - PdV + \sum_{i=1}^n \mu_i dN_i \quad n\text{-component mixture}$$

All thermodynamic properties can be obtained from the fundamental equation of a system. [see Ex. 5.1]

### Manipulation of Derivatives §5.2

For well-behaved mathematical functions, can we:

Inversion  $\left(\frac{\partial x}{\partial y}\right)_z = \frac{1}{\left(\frac{\partial y}{\partial x}\right)_z}$

Commutation  $\frac{\partial}{\partial z} \left(\frac{\partial x}{\partial x}\right)_z = \frac{\partial}{\partial y} \left(\frac{\partial x}{\partial z}\right) = \frac{\partial^2 x}{\partial x \partial z}$

XYZ-1 "Triple Product"  $\left(\frac{\partial x}{\partial y}\right)_z \left(\frac{\partial y}{\partial z}\right)_x \left(\frac{\partial z}{\partial x}\right)_y = -1$

Chain rule:  $\left(\frac{\partial x}{\partial y}\right)_z = \left(\frac{\partial x}{\partial w}\right)_z \left(\frac{\partial w}{\partial y}\right)_z = \frac{\left(\frac{\partial x}{\partial w}\right)_z}{\left(\frac{\partial y}{\partial w}\right)_z}$

These are functional relationships  $\left. \begin{array}{l} x = f(y, z) \\ \text{or } y = f(x, z) \\ \text{etc} \end{array} \right\}$

Euler's Theorem for Homogeneous Functions

$$f(\lambda x, \lambda y, z) = \lambda f(x, y, z) \quad \left. \begin{array}{l} \text{homog. of degree} \\ 1 \text{ with respect} \\ \text{to } x, y \text{ only} \end{array} \right\}$$

$$\text{then: } f = x \frac{\partial f}{\partial x} + y \frac{\partial f}{\partial y}$$

$$\text{e.g. } f = xz^2 + yz; \quad \frac{\partial f}{\partial x} = z^2 \quad \frac{\partial f}{\partial y} = z \quad \text{checks}$$

$$\text{Proof: } \frac{\partial f(\lambda x, \lambda y, z)}{\partial \lambda} = f = \frac{\partial f(\lambda x)}{\partial (\lambda x)} \frac{\partial (\lambda x)}{\partial \lambda} + \frac{\partial f(\lambda y)}{\partial (\lambda y)} \frac{\partial (\lambda y)}{\partial \lambda}$$

$$= x \frac{\partial f}{\partial x} + y \frac{\partial f}{\partial y} \quad \text{Q.E.D.}$$

Apply to  $u$ , which is homogeneous of degree 1 in all its variables  $u(\lambda S, \lambda V, \lambda N) = \lambda u(S, V, N)$  ( $u$  is extensive)

$$\therefore u = \left( \frac{\partial u}{\partial S} \right)_{V, N} S + \left( \frac{\partial u}{\partial V} \right)_{S, N} V + \left( \frac{\partial u}{\partial N} \right)_{V, S} N \Rightarrow$$

$$\Rightarrow \boxed{u = TS - PV + \mu N} \quad \text{Euler-integrated F.E.}$$

$u$  is still  $u(S, V, N)$ ; in this expression,  
 $T = T(S, V, N)$ ;  $P = P(S, V, N)$ ;  $\mu = \mu(S, V, N)$

Legendre Transformations

$(S, V, N)$  is not usually a convenient set of independent variables.

We often would like to work at  $(T, V, N)$   
 $(T, P, N)$

etc.  
 How do we obtain fundamental Equations in other variables?

In principle, from  $f(x)$  with  $\frac{df}{dx} = w$  we can obtain  $f(w)$ . But  $f(w)$  is not equivalent to  $f(x)$ !

$$\text{E.g. } f(x) = x^2 + 2 \Rightarrow w = \frac{df}{dx} = 2x \Rightarrow f(w) = \frac{w^2}{4} + 2$$

$$\Rightarrow w = \pm 2\sqrt{f-2} \quad (1)$$

From  $f(w)$ , inversion rule  $\frac{df}{dx} = w \Rightarrow \frac{dx}{df} = \frac{1}{w} \Rightarrow$

$$\Rightarrow x = \int \frac{1}{w} df \stackrel{(1)}{=} \int \frac{df}{\pm 2\sqrt{f-2}} = \pm \sqrt{f-2} + c \Rightarrow$$

$$f = (x-c)^2 + 2 \quad \text{not the same as starting } f(x)$$

Solution: define new function (transform)

$$g(w) = f - xw$$

$$dg = \underbrace{df}_{w dx} - xdw - wdx = -x dw$$

new function (first Legendre transform)

has the original derivative as variable

The original function  $f(x)$  can be obtained unambiguously from  $g(w)$  by differentiation:

$$dg = -x dw$$

$$\boxed{f = g + xw}$$